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## **SOME RESEARCH REGARDING THE DUCTILE IRONS ELABORATED IN ELECTRIC ARC FURNACES**

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### **ABSTRACT**

The present paper presents a series of experiments concerning the concocting of irons destined to the modification in electric arc furnaces, with basic wall lining. The chosen modifier is a classical one, based on Fe, Si, and Mg.

It will be presented the chemical compositions of irons that are subject to modification, the parameters of the modification process, and also the results (the final composition of ductile irons, the characteristics of the graphite and of the matrix, gases contents).

The conclusions can be used for the making up of a modifying technology, on an industrial scale. This is similar with the experimenting technology.

### **KEYWORDS:**

modification, ductile iron, graphite, magnesium.

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### **1. FOREWORD**

It is known that at the concocting of ductile irons, the electric arc furnaces are being used only at the making up of superior quality irons. When a low quality raw material is used, the electric arc furnace oven with basic wall lining is the only aggregate that allows the de-sulfuration and de-phosphoration processes.

The following experiments were made in an electric arc furnace with basic wall lining, with a 5 tones capacity, which was set on a steel casting platform. The resulting irons are destined to the nodulizing modification of the graphite.

### **2. EXPERIMENTS**

The materials from the oven were chosen so that they would satisfy low percentage of sulphur and other anti-modifier elements and in the same time high percentage of carbon in order to limit the carburating process. The weighted 4700...5500 kg, and was made from 85...90% new and old cast iron and from 10...15% old iron. Depending on the ingredients of the raw materials, the chemical composition of the resulting cast irons was presented in Table 1.

For the modification it was chosen the pot modifying technology together with the help of the modifying bell. The modifier quantity was calculated taking into account the sulphur percentage from the resulting cast irons as well as the parameters of the modifying process. The relationship is the following:

$$Q_{\text{mod}} = \frac{0,76(S - 0,10) + Mg_{\text{rez}} + t \cdot 10^{-3} \cdot \left(\frac{T}{1450}\right)^2}{\eta \cdot \frac{Mg_{\text{prealloy}}}{100}} \cdot G \quad (1)$$

where:

$Q_{\text{mod}}$  – the alloy quantity needed for modification;

$S$  – the sulphurus percentage in iron exposed to the modification, %

$Mg_{\text{rez}}$  – the proposed magnesium in iron, %

$t$  – the time required to maintain iron in, after modification, min

$\eta$  – the assimilation efficiency of the magnesium

$Mg_{\text{prealloy}}$  – the quantity of magnesium in prealloy,

$T$  – modification temperature, °C

$G$  – the quantity of modified iron, kg

The used prealloy had the chemical composition, which is presented in Table 2. The limits of molted irons chemical composition, before modification, were presented in Table 3. The residual elements in ductile iron are presented in Table 4.

Table 1. Chemical composition of the resulting cast irons

| Chemical composition. % |      |      |       |       |       |                   | Sc   | $\eta_{Mg}$ % |
|-------------------------|------|------|-------|-------|-------|-------------------|------|---------------|
| C                       | Si   | Mn   | P     | $S_i$ | $S_f$ | $Mg_{\text{rez}}$ |      |               |
| 3.68                    | 1.56 | 0.66 | 0.110 | 0.030 | 0.004 | 0.048             | 0.98 | 25            |
| 3.72                    | 1.73 | 0.70 | 0.110 | 0.022 | 0.002 | 0.043             | 1.01 | 12            |
| 3.60                    | 1.92 | 0.68 | 0.110 | 0.020 | 0.013 | 0.005             | 0.98 | -             |
| 3.64                    | 1.87 | 0.92 | 0.127 | 0.021 | 0.019 | -                 | 0.99 | -             |
| 3.80                    | 2.03 | 0.79 | 0.120 | 0.025 | 0.008 | 0.063             | 1.05 | 38            |
| 3.70                    | 1.34 | 0.76 | 0.095 | 0.021 | 0.011 | 0.059             | 0.96 | 28            |
| 3.72                    | 1.80 | 0.75 | 0.130 | 0.033 | 0.010 | 0.037             | 1.01 | 23            |
| 3.60                    | 1.94 | 0.79 | 0.150 | 0.030 | 0.010 | 0.080             | 0.99 | 40            |
| 3.66                    | 1.66 | 0.78 | 0.140 | 0.023 | 0.009 | 0.031             | 0.98 | 21            |
| 3.73                    | 1.95 | 0.68 | 0.113 | 0.022 | 0.011 | 0.048             | 1.02 | 28            |

Table 2. The chemical composition of used prealloy

| Mg, %       | Si          | Ca      | Fe        |
|-------------|-------------|---------|-----------|
| 10.0...15.0 | 40.0...45.0 | max 1.0 | remainder |

Table 3. The limits of molted irons chemical composition

| Limits of chemical composition, % |      |      |       |       | Sc   |
|-----------------------------------|------|------|-------|-------|------|
| C                                 | Si   | Mn   | P     | S     |      |
| 3.15                              | 1.34 | 0.68 | 0.095 | 0.020 | 0.98 |
| 3.70                              | 1.92 | 0.80 | 0.110 | 0.030 | 1.05 |

Table 4. Residual elements in cast irons

| Chemical composition, ppm |     |     |     |     |    |     |     |     |
|---------------------------|-----|-----|-----|-----|----|-----|-----|-----|
| Cr                        | Ni  | Mo  | Co  | Ti  | V  | Cu  | Pb  | As  |
| 11                        | 414 | 163 | 113 | 443 | 81 | 164 | 183 | 108 |
| 88                        | 310 | 17  | 113 | 198 | 26 | 178 | 45  | 44  |
| 58                        | 484 | 29  | 123 | 221 | 29 | 219 | 8   | 71  |
| 9                         | 251 | 76  | 76  | 139 | 37 | 125 | 63  | 52  |
| 125                       | 461 | 96  | 188 | 215 | 79 | 77  | 194 | 113 |
| 96                        | 284 | 22  | 104 | 147 | 24 | 163 | 44  | 40  |
| 100                       | 293 | 8   | 105 | 152 | 25 | 168 | 36  | 42  |
| 98                        | 288 | -   | 55  | 157 | 25 | 166 | 31  | 41  |
| 12                        | 440 | 189 | 112 | 299 | 90 | 201 | 210 | 119 |

The sequence of the modification operations of cast iron was the following:

- ✚ Molted iron, elaborated in the 5 tone furnace was melted in the 8 tone modification pot, after which the transportation of this on the casting platform was executed;
- ✚ In the modification area the molted iron was cleaned of the formed slag and the temperature was measured;
- ✚ After closing the modification pot-bell, the actual modification was undertaken. The reaction during modification was strong due to the high temperature of the iron. The modification time was 3-4 minutes;
- ✚ After cleaning the pot of the slag formed during the modification, the casting was executed. The stop time between modification and casting process was 6-8 minutes and the duration of modification was 1-2 minutes.

The total duration to maintain the molted status did not go over 12 minutes (see Table 5). The microstructures are shown in Table 6. The cast pieces out of the modified irons (ingots respectively) were exposed to a heat treatment for detension (see Table 7), according to the following pattern:

- ✚ heating from 150 to 650°C with a speed of 100°C/h, for 5 hours
- ✚ maintain at 650°C for 4 hours
- ✚ cool from 650°C to 300°C with a speed of 100°C/h
- ✚ free cooling from 300°C to the environmental temperature

Table 5. The parametres of the modification process

| No | Temperatures, °C |         | Durations, min |      |      |       |
|----|------------------|---------|----------------|------|------|-------|
|    | Modifying        | Casting | modif          | stay | cast | total |
| 1  | 1370             | 1300    | 3              | 7    | 2    | 12    |
| 2  | 1300             | 1250    | 2              | 8    | 1    | 11    |
| 3  | 1260             | 1200    | 2              | 6    | 1    | 9     |
| 4  | 1270             | 1190    | 2              | 7    | 1    | 10    |
| 5  | 1300             | 1240    | 3              | 7    | 2    | 12    |
| 6  | 1340             | 1280    | 2              | 8    | 1    | 11    |
| 7  | 1330             | 1260    | 2              | 6    | 2    | 10    |
| 8  | 1350             | 1260    | 2              | 6    | 2    | 10    |
| 9  | 1360             | 1310    | 3              | 7    | 1    | 11    |
| 10 | 1350             | 1280    | 3              | 7    | 1    | 11    |

  

| No | Prealloy    |          | Q <sub>iron</sub> , to |
|----|-------------|----------|------------------------|
|    | Type        | Quantity |                        |
| 1  | Fe-Si-Ca-Mg | 100      | 5.0                    |
| 2  | Fe-Si-Ca-Mg | 100      | 5.0                    |
| 3  | Fe-Si-Ca-Mg | 75       | 4.8                    |
| 4  | Fe-Si-Ca-Mg | 85       | 5.0                    |
| 5  | Fe-Si-Ca-Mg | 80       | 5.0                    |
| 6  | Fe-Si-Ca-Mg | 85       | 4.5                    |
| 7  | Fe-Si-Ca-Mg | 75       | 5.2                    |
| 8  | Fe-Si-Ca-Mg | 80       | 5.0                    |
| 9  | Fe-Si-Ca-Mg | 80       | 4.9                    |
| 10 | Fe-Si-Ca-Mg | 75       | 45.0                   |

Table 6. Microstructure of the ductile irons

| No | Microstructure              |                     |                       |                    |
|----|-----------------------------|---------------------|-----------------------|--------------------|
|    | Graphite                    | Surface occupied, % | Graphite diameter, μm | Matrix             |
| 1  | Nodular + vermicular        | 10.91               | 75                    | Ferrite            |
| 2  | Nodular                     | 9.52                | 60                    | Ferrite            |
| 3  | Flake                       | 14.50               | -                     | Ferrite            |
| 4  | Flake                       | 11.91               | -                     | Ferrite            |
| 5  | Nodular                     | 7.71                | 70                    | Ferrite            |
| 6  | Nodular                     | 10.09               | 75                    | Ferrite + pearlite |
| 7  | Nodular                     | 13.02               | 90                    | Ferrite + pearlite |
| 8  | Nodular + vermicular        | 11.44               | 62                    | Ferrite + pearlite |
| 9  | Nodular + flake             | 6.52                | 70                    | Ferrite + pearlite |
| 10 | Nodular, vermicular + flake | 11.84               | 70                    | Ferrite + pearlite |

Table 7. Heat treating

| No | Mg <sub>rez</sub> % | Heat treating |
|----|---------------------|---------------|
| 1  | 0,049               | Detension     |
| 2  | 0,043               | Detension     |
| 3  | 0,005               | Detension     |
| 4  | -                   | -             |
| 5  | 0,063               | Detension     |
| 6  | 0,059               | Detension     |
| 7  | 0,037               | Detension     |
| 8  | 0,080               | Detension     |
| 9  | 0,031               | Detension     |
| 10 | 0,048               | Detension     |

#### 4. CONCLUSIONS

- ✚ The elaboration of cast irons for modification in electric arc furnaces is possible especially due to the fact that the process for elaboration can be led much easier, resulting in less quantities of sulphurus, the main restrictive element in the composition of chemical composition, as well as other hampering elements (P, O, H, N).
- ✚ the temperatures at the moment of evacuation from the furnace can be more easily framed within the limits indicated by the specialized readings in modification process

- ✚ Due to the less quantities of hampering elements within the composition, the quantities of pre-alloy unmodifying required to ensure the modification effect of the graphite are smaller.
- ✚ The duration of the modification effect is longer and the graphite nodulizing is more secure
- ✚ The proportion of nodular graphite is bigger than in the case of ductile irons obtained in other types of furnaces, and all the characteristics of the graphite (size, distribution) are better than the previously mentioned process
- ✚ The proposed technology can be applied for the elaboration of ductile irons in electric arc furnaces of any size.

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